# JOURNAL OF THE CHEMICAL SOCIETY <br> ERRATA 

Vol. 1957, page 1902, line 12. For 1:7-dichloroanthraquinone read 1:7-dichlorodianthranilide.
Vol. 1957, page 4205 , line $14^{*}$. For $89^{\circ} / 0 \cdot \mathrm{~mm}$. read $89^{\circ} / 0 \cdot 3 \mathrm{~mm}$.
Vol. 1959, page 2568 , line 24. For the equation given read $A_{j}=\alpha C p_{C}\left[\varepsilon_{\mathrm{C}}-n p_{\mathrm{R}} \varepsilon_{\mathrm{R}}\right] /\left\{p_{\mathrm{R}}+\left(k_{\mathrm{R}} /\left[\mathrm{H}^{+}\right]\right)\right\}$.
Vol. 1960, page 232, line 6 *. For 1-dibromomethylnaphthalene read 2-dibromomethylnaphthalene.

Vol. 1960, page 604. Infrared absorption bands recorded, in this Paper, as observed below $400 \mathrm{~cm} .^{-1}$ are erroneous owing to incorrect use of the spectrophotometer, and the portions of Figs. 1, 3, and 4 intended to represent these spectra in the region $350-400 \mathrm{~cm} .^{-1}$ should be disregarded.

Vol. 1961, page 2942, line 3, should read From experiments (Swindale, Swedlund, and Robertson, J., 1950,812 ) involving addition of bromine to various allylic compounds, an order $\mathrm{F}>\mathrm{Cl}>\mathrm{Br}>\mathrm{CN}$ was obtained for the reactivity of X in $\mathrm{X} \cdot \mathrm{CH}_{2} \cdot \mathrm{CH}=\mathrm{CH}_{\mathbf{2}}$.

Vol. 1961, page 2942, line 6. For Baker et al. read Swindale et al.
Vol. 1961, page 4682, line 17. For 2-acetyl-1-hydroxy- and 1-acetyl-2-hydroxy-anthra- read 1-acetyl-2-hydroxy- and 2 -acetyl-1-hydroxy-anthra-.

Vol. 1961, page 4683, Table, lines 17 and 18. For 2-Methyl read 3-Methyl, and for 2-Phenyl read 3-Phenyl.

Vol. 1961, page 4928, Table. Rate constants for $\mathrm{CF}_{2} \cdot \mathrm{CO}_{2} \mathrm{H}-\mathrm{H}_{2} \mathrm{O}-\mathrm{H}_{2} \mathrm{SO}_{4}$ refer to $24 \cdot 8^{\circ}$ not $70 \cdot 1^{\circ}$, and figures listed as rate constants for $\mathrm{CF}_{3} \cdot \mathrm{CO}_{2} \mathrm{H}-\mathrm{H}_{2} \mathrm{O}-\mathrm{HClO}_{4}$ are rate factors at $\mathbf{2 5 \cdot 0} \mathbf{0}^{\circ}$.

Vol. 1961, page 5213, Table 2. For Diethyl $\alpha \alpha$-di-(2-cyanoethyl)acetoacetate (III) read Methyl $\alpha \alpha$-di-(2-cyanoethyl)acetoacetate (III).

Vol. 1961, page 5331, line 10. For The rates of recombination read The rate constants of recombination.
Vol. 1961, page 5331, line 12. For $8 \times 10^{7}$ and $3 \times 10^{5} 1$. mole ${ }^{-1}$ sec. $^{-1}$ read $8 \times 10^{13}$ and $3 \times 10^{11}$ 1. $\mathrm{mole}^{-1} \mathrm{sec} .^{-1}$.

Vol. 1961, page 5464 , line 15 . For 5.3 read 6.06 . Line 16 . For 6.5 read 7.26 .
Vol. 1962, page 175 , line 33 . For iodate ( 2 ml .) and 3 N -sulphuric acid ( 2 ml .) read iodide ( 2 ml .).
Vol. 1962, page 832, line $13^{*}$. For $\left(8.5 \times 10^{-5} 1 . \mathrm{mole}^{-1} \mathrm{sec}^{-1}\right)$ read $\left(3.7 \times 10^{-5} 1 . \mathrm{mole}^{-1} \mathrm{sec}^{-1}\right)$.
Vol. 1962, page 668, Title and throughout the paper. For Nitroso(dimethyldithiocarbonato)cobalt read Nitroso(dimethyldithiocarbamato)cobalt.

Vol. 1962, page 901, line 7. For Ethyl piperidine-2-carboxylate read ethyl 1-methylpiperidine-2carboxylate.

Vol. 1962, page 901, line 13. For 1-(piperidine-2-carbonyl)piperidine read 1-(1-methylpiperidine-2carbonyl) piperidine.

Vol. 1962, page 989. Ref. 11. For Vol. 33 read Vol. 25.
Vol. 1962, page 1519, line 23. For ( 1 ml .) read (11.).

* From bottom of main text.


## Errata．

Vol．1962，page 1828，Table 4，Column 3．For Slightly $>\mathrm{Me}$ ，and Slightly $>$ Et read Slightly $<\mathrm{Me}$ ， and Slightly＜Et．

Vol．1962，page 2206，Formula（III）should read as tricyclo［5，2，1， $\left.0^{2,6}\right]$ decane


Vol．1962，page 2210 ，line 3 ＊．For $\mathrm{C}_{44} \mathrm{H}_{12} \mathrm{NO}_{4} \operatorname{read} \mathrm{C}_{18} \mathrm{H}_{13} \mathrm{NO}_{4}$ ．
Vol．1962，page 2383，Table．Column heading＇Desilylation＇applies to first column only，and ＇Detritiation＇applies to the other three columns．

Vol．1962，page 2996，line $13^{*}$ ．For $\left(\mathrm{Me}^{\cdot} \cdot \mathrm{C}_{6} \mathrm{H}_{4} \cdot \mathrm{O}\right)_{2}$ read $\left(\mathrm{Me} \cdot \mathrm{C}_{6} \mathrm{H}_{4} \cdot \mathrm{O}\right)_{2} \mathrm{TiCl}_{2}$ ．
Vol．1962，page 3851，Table 2．Col． $7\left[S\left(\psi, 3 d_{x^{2}-y^{2}}\right)\right]$ should read－，一，一，一，一， $0.150(-2.0)$ ， $0.157(-2.8), 0.052(-4.5),-,-$ ，and then as in the published Table．The corresponding entries in col． 8 should be zero．

Vol．1962，page 3961，Ref．1．For Vol． 3 read Vol． 8.
Vol．1962，page 3962，Table，5th column．For H $\dagger$ read H＊．
Table，lst column，1lth entry．For $n-\mathrm{C}_{7} \mathrm{H}_{5}$ read $\mathrm{n}-\mathrm{C}_{5} \mathrm{H}_{11}$ ．
Vol．1962，page 3963，line 3．For sodium phosphite read sodium hypophosphite．
Vol．1962，page 4021；Scheme 1．Formula（X）should read $\left[\mathrm{ReOI}_{\mathbf{3}} \mathrm{L}_{2}\right]$ ．
Vol．1962，page 4065 ，line 3．For plot of $\log _{10} k$ against $\mu$ ，read plot of $\log _{10} k$ against $\mu^{\frac{1}{2}}$ ．
Vol．1962，page 4595，line 36．For 0.0 read 0.01 ．
Vol．1962，page 4807，Fig．2B，$y$－axis．For -50 to -250 read $-5 \cdot 0,-10 \cdot 0,-15 \cdot 0,-20 \cdot 0,-25 \cdot 0$ ．
Vol．1962，page 5110，line 21．For 4－oxo－1－phenylphosphacyclo－read 1－phenylphosphacyclo－hexan－ 4－one．

Vol．1962，page 5121，line 15．For 4＇，7－dimethoxyflavone read 4＇，7－dimethoxyflavanone；line 19．For 4＇，7－dimethoxyflavanone read $\mathbf{4}^{\prime}, 7$－dimethoxyflavone．
＊From bottom of main text．

